

# **Identification, Isolation and Characterization of a Novel** Glucosyltransferase from Beauveria bassiana Jie Ren, Jixun Zhan\*

## **1. Introduction**

Glycosylation is a widely occurring reaction in nature and one of the important reactions for the formation of various natural glucosides with different applications. Glycosylation is catalyzed by glycosyltransferases (GTs) in living organisms. GTs transfer sugar moieties from activated donor molecules to various acceptors. Family 1 glycosyltransferase (GT1) includes the uridine 5'diphosphate (UDP)-glycosyltransferases that are involved in the glycosylation of flavonoids. Thereinto, UDP-glucose is the most typical donor molecule for GT1 and the corresponding GTs are called UDP-glucosyltransferases (UGTs) (Fig. 1).



Fig.1 UGTs-mediated glucosylation process.

Flavonoids are polyphenolic natural products with diverse bioactivities, such as antibacterial, antimicrobial, antidiabetes, anti-inflammatory, antioxidant, antitumor, antiallergic, antihypertensive, and antiviral properties.

The most representative antioxidant flavonoid is quercetin, but its low aqueous solubility and poor bioavailability hinder further clinical applications. Here, we report a novel GT from *Beauveria bassiana* that can be used to prepare quercetin-3-O- $\beta$ -D-glucoside and quercetin-7-O- $\beta$ -D-glucoside (Fig. 2).



Fig.2 Enzymatic production of glucosylated quercetin

## 2. Hypothesis

Fungi are well-known for the production of various bioactive compounds. My hypothesis is that these strains are rich in natural product biosynthetic enzymes, some of which may be used for preparation of glycosides. Therefore, our research team is interested in discovering useful glycosyltransferases from microbes.

162 Da. Therefore, this product was characterized as glucosylated

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quercetin (Fig. 4).

Compound 3 was glycosylated at C-7, which was confirmed by the HMBC correlation of H-1" ( $\delta$  5.07, d, J=7.5 Hz) to C-7 at  $\delta$ 162.7. Meanwhile, compound 4 was glycosylated at C-3, which was confirmed by the HMBC correlation of H-1" ( $\delta$  5.46, d, J=7.5 Hz) to C-3 at  $\delta$  133.3 (Fig. 6)

Fig. 6 Chemical structures of glucosylated quercetin. a Chemical structure of compound 3 in Quecetin+E. coli BL21(DE3)/pET28a-Bbgt; b Chemical structure of compound 4 in Quecetin+S. cerevisiae BJ5464/pET28a-Bbgt.







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# Conclusion

This research was focused on the discovery of the novel glucosyltransferase from Beauveria bassiana ATCC 7159, which provides a novel tool for the preparation of water-soluble derivatives of medicinally important natural product. This enzyme can be used to prepare different glucosides with different expression hosts. Meanwhile, different sugar-acceptors can be glucosylated at different chemical structure positions with different sugar moieties. Therefore, the BbGT enzyme is expected to be useful molecular tool for increasing the bioavailability of various healthbeneficial natural products, so that the bioactive components with low bioavailability can be better utilized by the human body.





### **3.5 Determination of the optimal reaction conditions of BbGT**

The optimal catalyzing temperature and pH conditions are 35°C and 8.0 respectively. And the BbGT activity was stimulated by Ca<sup>2+</sup>, Mg<sup>2+</sup>, and Mn<sup>2+</sup>, and when Zn<sup>2+</sup> was added to the reaction system, we found that it showed strong inhibition to this enzyme (Fig. 7).



Fig. 7 Determination of the optimum reaction conditions of BbGT. a Effect of reaction temperature on the BbGT glucosylation activity. **b** Effect of reaction pH on the BbGT glucosylation activity. **c** Effect of reaction various metal ions on the BbGT glucosylation activity. Data are presented as the mean  $\pm$  SD from three independent experiments.

### **Future Work**

nize the *in vivo* production conditions of quercetin osides using different hosts expressing BbGT. uction of quercetin glycosides by repeated uses of E. coli BL21(DE3)/pET28a-BbGT in 1L batch reaction system.