# Photochemistry of wildfire derived pyrogenic-dissolved organic matter (py-DOM) for photodegradation of aquatic contaminants

Presenter: Monika Madhiyan, Principal Advisor: Dr. Kyle Moor Department of Civil and Environmental Engineering | Utah State University | Utah Water Research Lab | Moor Research Group

### Introduction

Pyrogenic dissolved organic matter (py-DOM) is a type of chromophoric (or light absorbing) dissolved organic matter (CDOM) that can generate, in presence of sunlight or similar light sources, photochemically produced reactive intermediates (PPRI) such as singlet oxygen ( ${}^{1}O_{2}$ ), the excited form of ground-state oxygen (Fig.1.a). <sup>1</sup>O<sub>2</sub> is an important oxidant during the phototransformation of pollutants in water and impacts their lifetimes. Currently changing wildfire intensity necessitate and frequency understanding the role of py-DOM in the production of <sup>1</sup>O<sub>2</sub> and therefore in pollutant transformation.  ${}^{1}O_{2}$  quantum yield ( $\Phi_{\Lambda}$ , number of <sup>1</sup>O<sub>2</sub> molecules generated upon absorption of one photon of light) for pyDOM extracted from laboratory prepared chars has been reported in previous studies, but these may inaccurately represent actual wildfire chars present in the environment.

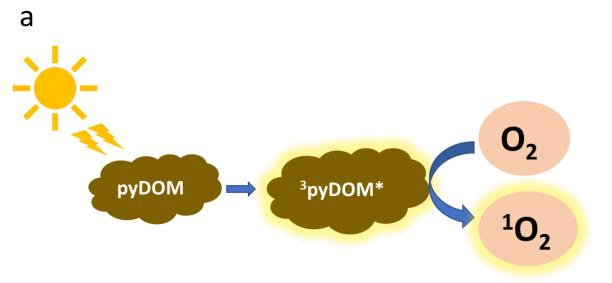




Fig 1. (a) Photochemistry of pyDOM, (b) Colorado Grizzly Creek wildfire site

In this study, py-DOM was extracted from wildfire-derived chars and their  $\Phi_{\Lambda}$  at 365 nm was determined by measuring the time-resolved <sup>1</sup>O<sub>2</sub> phosphorescence at 1275 nm. This determination method was validated measuring  $\Phi_{\Lambda}$  of SRNOM (Suwanee River Natural Organic Matter), which has been established in previous studies (1.8±0.2%) that use the same  $\Phi_{\Lambda}$ determination method as here.

## Methods

Biochar solids were obtained by physically scraping burnt log samples collected from the Colorado Grizzly Creek (CGC) wildfire site (Fig.1.b). Additionally, store-bought Oak wood pieces were burnt in a muffle furnace at 250 °C for 3 hours. These solids were ground and sieved to ensure maximum homogenization and then agitated in 10 mM phosphate buffer (pH=7.0) for 48 hours followed by centrifugation and filtration to extract py-DOM (Fig.2). <sup>1</sup>O<sub>2</sub> phosphorescence signals from py-DOM and reference samples were obtained by using a tunable excitation wavelength (365 and an automated spectrophotometer coupled with a NIR-PMT detector (Fig. 3).  $\Phi_{\Lambda}$  for pyDOM and SRNOM was calculated relative to a known  $\Phi_{\Lambda}$  value of the reference, Perinaphthenone (95%) (Eqn. 1).





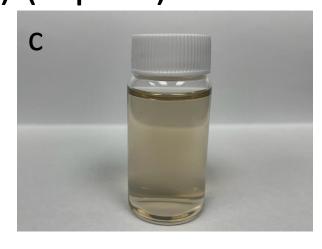


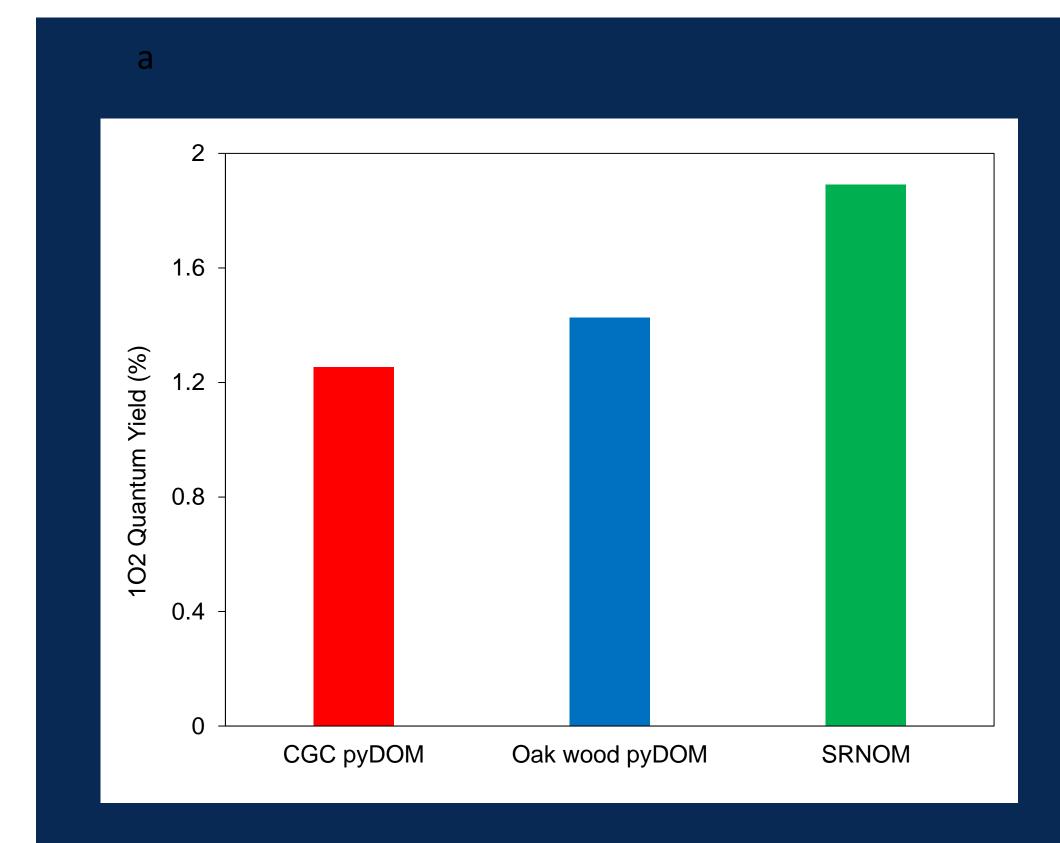
Figure 2. (a) Collected sample portion. (b) Chipped off pieces from the sample. (c) Obtained pyDOM solution after grinding, sieving, agitation, centrifugation and filtration.

# —Oak wood pyDOM

**Figure 3.** Time-resolved <sup>1</sup>O<sub>2</sub> phosphorescence kinetic trace measured at 1275 nm. Insert: UV-Vis absorption spectra for prepared pyDOM and SRNOM

## Results and Discussion

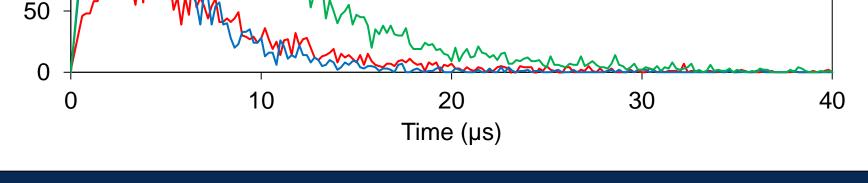
- Extracted pyDOM solutions from CGC wildfire and Oak wood biochar produced <sup>1</sup>O<sub>2</sub> phosphorescence signals (Fig. 3).
- $\Phi_{\Lambda}$  at 365 nm for CGC py-DOM and storebought Oak wood pyDOM was found to be 1.25% and 1.43% respectively (Fig. 4). The difference between these quantum yields can be attributed to the different chromophoric moieties present in the pyDOM which determine their absorption of light and their rate of reaction with ground state O<sub>2</sub> to generate  ${}^{1}O_{2}$ .
- The determined  $\Phi_{\Lambda}$  for both wildfire derived and lab-prepared pyDOM is lower than other studies that have reported  $\Phi_{\Lambda}$  for lab prepared chars. This could be due to different char and/or the different chemistry measurement technique.
- $\Phi_{\Lambda}$  for SRNOM at 365nm was found to be 1.9% which is consistent with previous findings and offered as a validation of this method.



**Figure 4.** Determined  $\Phi_{\wedge}$  for CGC Wildfire pyDOM, Oak wood pyDOM, and SRNOM at 365 nm.

## Conclusions and Future Work

- <sup>1</sup>O<sub>2</sub> was produced by wildfire-derived pyDOM  $(\Phi_{\Delta_{365nm}} = 1.25\%)$  and store-bought Oak wood pyDOM ( $\Phi_{\Delta_{365nm}} = 1.43\%$ ), which indicates that pyDOM from wildfire biochar can possibly play an important role in the photo-transformation and lifetime of pollutants present water.
- Future work involves addressing the discrepancy between the determined  $\Phi_{\Lambda}$  in this study and those reported in previous studies by comparing char chemistry and <sup>1</sup>O<sub>2</sub> measurement techniques.
- Wavelength-dependent  $\Phi_{\Lambda}$  for pyDOM and characterization of pyDOM is necessary to understand their <sup>1</sup>O<sub>2</sub> production under natural sunlit environment and light-involving water treatment systems and their different <sup>1</sup>O<sub>2</sub> production rates



 $I_{\lambda}$ , rel = Relative emission intensity of the laser measured over 1 nm wavelength (λ) intervals (counts)  $\alpha_{\lambda}$  = Measured decadic absorbance coefficient of the solution at  $\lambda$ l = Pathlength (1 cm)

 $R_{abs}$  = Rate of light absorbance (mmol photons cm<sup>-3</sup> s<sup>-1</sup>) given as,

Sens. = Sensitizer (Here, CGC wildfire pyDOM/ Oak wood pyDOM/ SRNOM)

 $R_{abs} = \frac{\Sigma_{\lambda} I_{\lambda}, rel(1-10^{-\alpha_{\lambda}l})}{l}$  (Equation 1.1)

Coll. t = Amount of time the signal was collected (s) (90 seconds for Perinaphthenone, 600 seconds for pyDOM and SRNOM)

Ref. = Reference (Here, Perinaphthenone)

 $s_{\Lambda}$  = Observed area of the  ${}^{1}O_{2}$  signal

 $\Phi_{\Lambda} = {}^{1}O_{2}$  quantum yield (%)

Equation 1.



